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New proton-conductive membranes for fuel cells based on hybrid composites

Malahova E.A.^a, Chernigovskaya M.A.^a, Raskulova T.V.^{a*}^aAngarsk State Technical Academy, Tchaikovsky, 60, Angarsk 665835, Russian Federation

Abstract

New hybrid organic-inorganic composites based on sulfur-containing styrene copolymers and allyl glycidyl ether and tetraethoxysilane were obtained by a sol-gel synthesis method. The membranes created on the basis of composites have proton-conductive properties and are characterized by a higher heat-exchange capacity compared with commercial membranes as Nafion and MF-4 SK, enabling to consider them as perspective membrane materials for fuel cells.

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1. Introduction

In present days the problem of a global energy crisis related with the emission of a huge amount of toxic wastes as a result of energy resources traditional using aggravated in industrialized countries. It affects negatively the environment and causes irreversible climatic changes. The most urgent problem is searching for an alternative way of power production. One of the solutions of this problem is the development of fuel cells based on a polymeric proton-conductive membrane [1].

According to the analysis of studies on polymeric proton-conductive electrolytic membranes developing there are three groups of polymeric systems suitable for the efficient industrial application [2]:

- perfluorinated sulfur-containing polymers;
- condensation polymeric systems;

* Corresponding author. Tel.: +7-902-514-9351; fax: +7-395-551-29037.

E-mail address: raskulova@list.ru

- hybrid polymeric systems (hydrogels and composites).

In industrial scale, membranes based on perfluorinated sulfur-containing polymers are generally used. They are high tech and characterized by high performance values, but they can only be used at temperatures below 90 °C and, the most important thing is the high cost, because of their production complexity. For example, Nafion membranes consist of a tetrafluorethylene copolymer and a perfluorinated sulfur-containing monomer [3]. Its producing process involves at least five independent technological stages.

Proton-conductive membranes based on condensation systems are, as a rule, formed on the basis of aromatic or heterocyclic high-molecular compounds, for example, based on polybenzimidazole, doped by the phosphoric acid. The advantage of such membranes is a high proton conductivity, considerable heat resistance (up to 600 °C), tolerance to CO and CO₂ impurities, low permeability for different types of fuel, including methanol, high mechanical strength [4]. However membranes based on those polymers are produced now as a pilot project because of the lack of commercial processes of the initial monomers production.

A new promising type of proton-conductive membranes is hybrid composite membranes formed by the method of sol-gel synthesis based on organic polymers and organic-silicon precursors [5]. The introduction of an organic-silicon framework in a polymeric matrix enables to improve significantly mechanical properties and thermal stability of the proton-conductive membranes [6]. An organic component, as a rule, contains functional groups providing a proton transport. The advantage of hybrid composite membranes is the possibility of their obtaining based on organic and hetero-organic synthesis products, which are simple in production technology and cheaper in cost than traditional ones. For example, for synthesis of such membranes styrene copolymer (St) and unsaturated epoxides can be used as an organic polymeric component and a tetraethoxysilane (TEOS) is used as the organic-silicon component.

The purpose of this work was to obtain hybrid organic-inorganic composite materials based on modified St and allyl glycidyl ether (AGE) copolymers using TEOS as an organic-silicon precursor by sol-gel synthesis method, and to study the main characteristics of polymeric membranes obtained on their basis.

2. Experimental

St and AGE copolymers were obtained by a suspension free radical copolymerization method according to [7]. To provide a proton conductivity, copolymers were exposed to preliminary sulfonation by the concentrated sulfuric acid ($\rho = 1,825 \text{ g/cm}^3$) in a benzene or toluene solution at the temperature of 60 to 90° C for 1-12 hours. The methods of turbidimetric titration, element analysis (C, H, S weight content), NMR ¹³C spectroscopy were used to determine the composition and structure of the obtained products. The formation of hybrid composites based on sulfonated copolymers was carried out similarly by sol-gel synthesis method [8]. The composites' structure was established according to the element analysis (S, N, Si weight content) and infrared spectroscopy. The research of the synthesized samples surface structure was conducted by the scanning electron microscopy method. The impedance spectroscopy method was used to characterize the membranes proton conductivity.

3. Methods

Turbidimetric titration was carried out at the temperature of 25° C, the initial copolymers solution concentration in cyclohexanone was 0,96 g/100 ml, isopropanol was used as a precipitator. The optical density of solutions was determined by photoelectrocolorimeter KFO-1. The element analysis of copolymers and composites was carried out by «Thermo Finnigan Flash EA 1112 Series» gas analyzer. Infra-red spectrums of composites were obtained in KBr tablets and in a microlayer using Specord IR-75 spectrometer, and Bruker IFS-25 spectrometer. NMR ¹³C spectrums of copolymers were registered on VXR– 500S spectrometer “Varian” (with 125,5 MHz operating frequency) with a noise isolation from protons and with a 2,5 sec relaxation delay. The impulse was 90°. Solvents were DMSO–d₆ and acetone–d₆. Chrome tris-acetylacetonate (0,02 mol/l) was used as a relaxant. The examination of a surface structure of synthesized samples was conducted by the scanning electronic microscopy method using “Philips-525-M apparatus”. Films electroconductivity was measured by a complex impedance technique in isothermal mode in 500000-5000 Hz frequency range using «Z-500PX» impedancemeter (Elins, Russia).

The main characteristics of the obtained membranes compared to the most widespread commercial membranes like Nafion (for example, Nafion 212) and MF-4 SK-1 are shown in table 2.

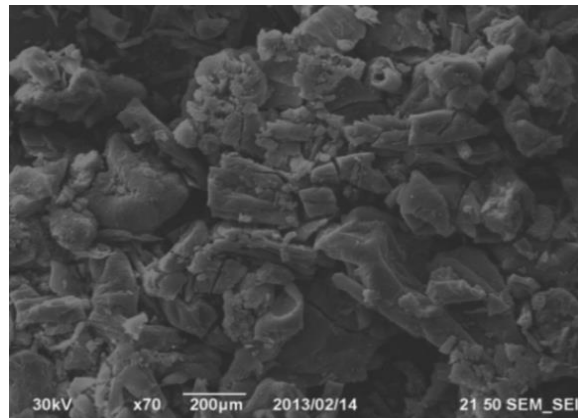


Fig. 1. Electronic microphotography of a composite surface.

Table 1. Measurement results of a hybrid membrane ionic conductivity (S 17,13% mass.).

№	t, °C	$(1/T) \cdot 10^3, K^{-1}$	S, cm ²	d, cm	R, Ohm	σ , S/cm	$\ln \sigma$	E _A , kJ/mol
1	30	3,30	0,139	0,014	227,50	$4,4 \cdot 10^{-4}$	-7,72	51,11
2	40	3,19	0,139	0,014	97,50	$2,0 \cdot 10^{-3}$	-6,88	
3	50	3,10	0,139	0,014	62,50	$1,6 \cdot 10^{-3}$	-6,43	
4	60	3,00	0,139	0,014	36,00	$2,8 \cdot 10^{-3}$	-5,88	

Table 2. Characteristics of proton-conductive composite membranes.

Name of characteristic	Value of characteristic		
	Nafion 212 membrane	MF-4 CK-1 membrane	Hybrid membranes
Specific conductivity, S/cm	$1,8 \cdot 10^{-2}$	$1,4 \cdot 10^{-2}$	$2 \cdot 10^{-3}$
Activation energy, kJ/mol	17,04	29,9	49,6-51,1
Exchange capacity, mg-eq/g	0,95-1,01	0,90-1,15	1,84-2,70
Modulus of elasticity, MPa	160,0	132,0	15,6

The comparison of the specified parameters shows that the samples of commercial membranes have higher values of the specific conductivity, smaller values of the activation energy and better mechanical properties. However hybrid membranes show a higher exchange capacity compared to commercial membranes (1,84-2,70 and 0,95-1,01 mg-eq/g, respectively) providing the opportunity to consider them as perspective membrane materials for fuel cells.

5. Conclusion

Thus, in the course of the conducted investigations:

- hybrid composite materials with a high elasticity, thermal and chemical resistance and proton conductivity of $1,46 \cdot 10^{-3}$ S/cm were obtained by the method of hydrolytic condensation of a tetraethoxysilane and sulfonated copolymers of styrene and allyl glycidyl ether;
- a high exchange capacity of the obtained composite membranes compared to commercial membranes as Nafion and MF-4 CK (1,84-2,70 and 0,95-1,01 mg eq/g, respectively) enables to consider them as promising membrane materials for fuel cells.

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